FORMATION AND STABILITY OF RADIATION PRODUCTS IN EUROPA'S ICY SHELL M. H. Moore<sup>1</sup>, R. L. Hudson<sup>2</sup>, R. W. Carlson<sup>3</sup>, and R. F. Ferrante<sup>4</sup> (<sup>1</sup>NASA/Goddard Space Flight Center, Greenbelt, MD 20771; Marla.H.Moore@nasa.gov, <sup>2</sup>Eckerd College, St Petersburg, FL 33733; hudsonrl@eckerd.edu, <sup>3</sup>JPL, Pasadena, CA 91109; tassie@earthlink.net, US Naval Academy, Annapolis, MD 21402; ferrante@usna.edu)

**Introduction:** Spectra of Europa reveal a surface dominated by water-ice [1] along with hydrated materials [2,3] and minor amounts of  $SO_2$  [4,5],  $CO_2$ [6], and  $H_2O_2$  [7]. Jovian magnetospheric ions (protons, sulfur, and oxygen) and electrons produce significant chemical modifications of the surface on time scales of a few years at micrometer depths [8]. Our laboratory studies examine the formation and stability of radiation products in H<sub>2</sub>O-rich ices relevant to Europa. Infrared (IR) spectra of ices before and after irradiation reveal the radiation destruction of molecules and the formation of products at 86 – 132 K. In addition, spectra of ices during warming track thermal evolution due to chemical changes and sublimation processes.

IR-identified radiation products in 86 - 132 K irradiated H<sub>2</sub>O + SO<sub>2</sub> ices are the bisulfate ion, HSO<sub>4</sub><sup>-</sup>, sulfate ion, SO<sub>4</sub><sup>2-</sup> and the hydronium ion, H<sub>3</sub>O<sup>+</sup>. Warming results in the formation of a residual spectrum similar to liquid sulfuric acid, H<sub>2</sub>SO<sub>4</sub>, for H<sub>2</sub>O:SO<sub>2</sub> ratios of 3:1, whereas hydrated sulfuric acid, H<sub>2</sub>SO<sub>4</sub> 4 H<sub>2</sub>O, forms for ratios of 30:1. Radiation products identified for irradiated H<sub>2</sub>O + H<sub>2</sub>S ices at 86 K are H<sub>2</sub>S<sub>2</sub> and SO<sub>2</sub>. When irradiated at 110 and 132 K, ices with H<sub>2</sub>O:H<sub>2</sub>S ratios of either 3:1 or 30:1 show the formation of H<sub>2</sub>SO<sub>4</sub> 4 H<sub>2</sub>O on warming to 175 K. We have also examined the radiation stability of H<sub>2</sub>SO<sub>4</sub>.

Addition of  $CO_2$  to  $H_2O + SO_2$  ices results in the formation of  $CO_3$  at 2046 cm<sup>-1</sup> (4.89 m). This is the strongest band from a carbon-containing product in the mid-IR spectral region, and it is also seen when either pure  $CO_2$  or  $H_2O + CO_2$  ice is irradiated. Experiments with  $CH_4$  added to  $H_2O + SO_2 + CO_2$  ices

addressed the question of methane's use as a marker of methanogens in an irradiated ice environment.

New results on the near-IR spectrum of pure  $H_2O_2$  will be included in this presentation. Interpretations of near-IR water bands, with  $H_2O_2$  present, will be discussed. Irradiations of  $H_2O_2$  and  $H_2O + H_2O_2$  mixtures, to examine the possibility of  $O_2$  and  $O_3$  formation [9], are currently under investigation and new results will be discussed.

These laboratory studies provide fundamental information on likely processes affecting the outer layer of Europa's icy shell. This layer has the coldest, most heavily bombarded material in which radiation chemical markers of  $H_2O_2$  and  $O_2$  have already been detected. Through downward mixing, gardening, and subduction, these surface species may become available for subsurface activity.

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